

# CH3H2-15 Electrochemistry and Interfaces

**26/27**

**Department**

Chemistry

**Level**

Undergraduate Level 3

**Module leader**

Julie Macpherson

**Credit value**

15

**Module duration**

5 weeks

**Assessment**

20% coursework, 80% exam

**Study location**

University of Warwick main campus, Coventry

---

## Description

### Introductory description

This module will describe the state-of-the-art in interfacial chemistry and electrochemistry; both are key to advances in nanotechnology, energy conversion, electrosynthesis, pollutant removal from water systems etc. Many of the concepts covered in this course are used practically in industrial and technology companies cutting across energy, environmental, pharma and healthcare.

### Module aims

The module will introduce the student to the importance of the interface in (electro)chemical processes of importance. This module as well as providing the fundamentals, will highlight wide ranging key applications of societal importance.

### Outline syllabus

This is an indicative module outline only to give an indication of the sort of topics that may be covered. Actual sessions held may differ.

The module is split into 4 distinct topics:

Topic 1: Introduction to the Solid and Liquid Interface

Topic 2: Surface Characterization Techniques (structural and chemical) including microscopy and surface sensitive spectroscopy

Topic 3: Introduction to the Solid – Electrolyte Interface and Dynamic Electrochemical Measurements

Topic 4: Electrochemistry in Action

## Learning outcomes

By the end of the module, students should be able to:

- Gain an appreciation of the importance of interfacial processes across chemistry
- Investigate interfacial chemistry and structure across the length scales from the nanoscale to industry scale
- Gain an appreciation of fundamentals of the electrode – electrolyte interface and introduction to dynamic electrochemistry
- Gain an appreciation of application of dynamic electrochemistry in electrocatalysis, fuel cells, electrosynthesis, environmental pollutant removal, biological and environmental species monitoring
- Exposure to the potential uses of AI

## Indicative reading list

[Reading lists can be found in Talis](#)

## Subject specific skills

Kinetics

Spectroscopy and Structural Characterisation

Experimental Chemistry

## Transferable skills

Problem solving

Sourcing information

Understanding of the scientific method

---

## Study

### Study time

Type	Required
Lectures	22 sessions of 1 hour (15%)
Total	150 hours

Type	Required
Practical classes	3 sessions of 1 hour (2%)
Private study	125 hours (83%)
Total	150 hours

## Private study description

N/A

## Costs

No further costs have been identified for this module.

---

## Assessment

You do not need to pass all assessment components to pass the module.

### Assessment group D1

Assessment component	Weighting	Study time	Eligible for self-certification
Group presentation	20%		No
<p>Students will work in teams to design a 4 - 5 day lab experiment suitable for use in the Year 3 labs. A list of potential topics related to the course content will be provided to the students, and each team must design an experiment based around one of these.</p> <p>Initial planning:</p> <ul style="list-style-type: none"> <li>• Students will be encouraged to make use of AI (Chat-GPT or similar) to develop a rough outline for the experiment (students will be required to submit the query and response generated as part of their final report)</li> <li>• The teams need to critically analyse this plan to identify any weaknesses in the proposed experimental methodology/analysis</li> <li>• 5 slide presentation, each slide based around one of the points above and presented by a different student.</li> </ul>			
Written Reassessment of Undergraduate Laboratory			No

	<b>Weighting</b>	<b>Study time</b>	<b>Eligible for self-certification</b>
Experiment			
single assessment - based on original submitted work, how could the experiment be improved, based on original feedback, to submit a 2 page of A4 report			

**Assessment component**

Centrally-timetabled examination (On-campus)	80%		No
Written Examination			

- 
- Answerbook Pink (12 page)
  - Graph paper
  - Periodic Tables
  - Students may use a calculator

Reassessment component is the same

### **Feedback on assessment**

Cohort level examination feedback provided via Moodle following the Exam Board.

[Past exam papers for CH3H2](#)

## **Availability**

### **Courses**

This module is Optional for:

- UCHA-4 Undergraduate Chemistry (with Intercolated Year) Variants
  - Year 4 of F101 Chemistry (with Intercolated Year)
  - Year 4 of F122 Chemistry with Medicinal Chemistry (with Intercolated Year)
- UCHA-3 Undergraduate Chemistry 3 Year Variants
  - Year 3 of F100 Chemistry
  - Year 3 of F121 Chemistry with Medicinal Chemistry
- Year 4 of UCHA-F107 Undergraduate Master of Chemistry (with Intercolated Year)
- UCHA-F109 Undergraduate Master of Chemistry (with International Placement)
  - Year 3 of F109 MChem Chemistry (with International Placement)

- Year 3 of F111 MChem Chemistry with Medicinal Chemistry (with International Placement)
- UCHA-4M Undergraduate Master of Chemistry Variants
  - Year 3 of F105 Chemistry
  - Year 3 of F125 MChem Chemistry with Medicinal Chemistry
- Year 4 of UCHA-F127 Undergraduate Master of Chemistry with Medicinal Chemistry (with Intercalated Year)